



# **Material Safety Data Sheet**

NFPA	HMIS	Personal Protective Equipment
200	Health Hazard 2 Fire Hazard 0	
	Reactivity	See Section 15.

Section 1. Chemic	al Product and Company Identification	Page Number: 1	
Common Name/ Trade Name	Sodium Acetate, 3 M Sterile Solution, pH 5.2	Catalog Number(s).	\$3065
	<u> </u>	CAS#	Mixture.
Manufacturer	SPECTRUM LABORATORY PRODUCTS INC.	RTECS	Not applicable.
	14422 S. SAN PEDRO STREET GARDENA, CA 90248	TSCA	TSCA 8(b) inventory: Water, Acetic acid, glacial
Commercial Name(s)	Not available.	CI#	Not applicable.
Synonym	Not available.	DICACEOEI	
Chemical Name	Not applicable.	IN CASE OF E	24hr) 800-424-9300
Chemical Family	Alkaline metal salt. (Contains Li, Na, K, Rb, Cs or Fr) (Salt.)	CALL (310) 516	6-8000
Chemical Formula	Not applicable.		
Supplier	SPECTRUM LABORATORY PRODUCTS INC. 14422 S. SAN PEDRO STREET GARDENA, CA 90248		

Section 2.Composition and Information on Ingredients					
			Exposure Limits		
Name	CAS#	TWA (mg/m³)	STEL (mg/m³)	CEIL (mg/m³)	% by Weight
1) Water 2) Sodium acetate trihydrate 3) Acetic acid, glacial	7732-18-5 6131-90-4 64-19-7	10	15		55-65 30-40 4-6

Toxicological Data Acetic acid, glacial: on Ingredients ORAL (LD50):

ORAL (LD50): Acute: 3310 mg/kg [Rat]. 4960 mg/kg [Mouse]. 3530 mg/kg [Rat].

DERMÀL (LD50): Acute: 1060 mg/kg [Rabbit]. VAPOR (LC50): Acute: 5620 ppm 1 hours [Mouse].

### Section 3. Hazards Identification

Potential Acute Health Effects

Hazardous in case of skin contact (irritant), of eye contact (irritant). Slightly hazardous in case of ingestion, of inhalation (lung irritant).

**Potential Chronic Health Effects** 

**CARCINOGENIC EFFECTS**: Not available.

MUTAGENIC EFFECTS: Mutagenic for bacteria and/or yeast. [Acetic acid, glacial].

TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available.

The substance may be toxic to kidneys, upper respiratory tract, skin, teeth.

Repeated or prolonged exposure to the substance can produce target organs damage.

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Section 4. First Aid	Measures
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water. Cover the irritated skin with an emollient. Remove contaminated dothing and shoes. Cold water may be used. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.
Serious Skin Contact	Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.
Inhalation	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.
Serious Inhalation	Not available.
Ingestion	Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention if symptoms appear.
Serious Ingestion	Not available.

Section 5. Fire and Explosion Data			
Flammability of the Product	Non-flammable.		
<b>Auto-Ignition Temperature</b>	Not applicable.		
<b>Flash Points</b>	Not applicable.		
Flammable Limits	Not applicable.		
<b>Products of Combustion</b>	Not available.		
Fire Hazards in Presence of Various Substances	Not applicable.		
Explosion Hazards in Presence of Various Substances	Non-explosive in presence of open flames and sparks, of shocks.		
Fire Fighting Media and Instructions	Not applicable.		
Special Remarks on Fire Hazards	Not available.		
Special Remarks on Explosion Hazards	Acetic acid vapors may form explosive mixtures with air.  Reactions between acetic acid and the following materials are potentially explosive: 5-azidotetrazole, bromine pentafluoride, chromium trioxide, hydrogen peroxide, potassium permanganate, sodium peroxide, and phorphorus trichloride.  Dilute acetic acid and dilute hydrogen can undergo an exothermic reaction if heated, forming peracetic acid which is explosive at 110 degrees C.  Reaction between chlorine trifluoride and acetic acid is very violent, sometimes explosive. (Acetic acid, glacial)		

Section 6. Acci	Section 6. Accidental Release Measures		
Small Spill	Dilute with water and mop up, or absorb with an inert dry material and place in an appropriate waste disposal container. If necessary: <b>Neutralize the residue with a dilute solution of sodium carbonate.</b> Finish cleaning by spreading water on the contaminated surface and dispose of according to local and regional authority requirements.		
Large Spill	Absorb with an inert material and put the spilled material in an appropriate waste disposal. <b>Neutralize the residue</b> with a dilute solution of sodium carbonate. Finish cleaning by spreading water on the contaminated surface and allow to evacuate through the sanitary system. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.		

Section 7. Handling and Storage	
Precautions	Do not ingest. Do not breathe gas/fumes/ vapor/spray. Never add water to this product. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents.
Storage	Keep container tightly closed. Keep container in a cool, well-ventilated area.

Section 8. Exposure C	Controls/Personal Protection
<b>Engineering Controls</b>	Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value.
Personal Protection	Splash goggles. Lab coat. Gloves Respiratory protection is not necessary for normal handling. Adequate general (room) ventilation or local exhaust (fume hood) is sufficient. Use a vapor respirator under conditions where exposure to the substance is apparent (e.g. generation of high concentrations of mist or vapor, inadequate ventilation, development of respiratory tract irritation), and engineering controls are not feasible. Be sure to use an approved/certified respirator or equivalent.
Personal Protection in Case of a Large Spill	Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.
Exposure Limits	Acetic acid, glacial  TWA: 10 STEL: 15 (ppm) [Australia]  TWA: 25 STEL: 27 (mg/m³) [Australia]  TWA: 10 STEL: 15 (ppm) from NIOSH  TWA: 25 STEL: 37 (mg/m³) from NIOSH  TWA: 25 STEL: 37 (mg/m³) from NIOSH  TWA: 10 STEL: 15 (ppm) [Canada]  TWA: 26 STEL: 39 (mg/m³) [Canada]  TWA: 25 STEL: 37 (mg/m³)  TWA: 10 STEL: 15 (ppm) from ACGIH (TLV) [United States] [1999]  TWA: 10 (ppm) from OSHA (PEL) [United States]  TWA: 25 (mg/m³) from OSHA (PEL) [United States]

Section 9. Physical and Chemical Properties			
Physical state and appearance	Liquid.	Odor	Acetic acid.
Molecular Weight	Not applicable.	Taste	Not available.
pH (1% soln/water)	Not available	Color	Clear Colorless
<b>Boiling Point</b>	The lowest known value is 100℃ (212℉) (Water). We ighted average: 101.39℃ (214.5℉)		
Melting Point	May start to solidify at 16.6°C (61.9°F) based on d ata for. Acetic acid, glacial.		
Critical Temperature	The lowest known value is 321.67℃ (611℉) (Acetic acid, glacial).		
Specific Gravity	Weighted average: 1.12 (Water = 1)		
Vapor Pressure	The highest known value is 2.3 kPa (@ 20°C) (Water) . Weighted average: 2.24 kPa (@ 20°C)		
Vapor Density	The highest known value is 2.07 (Air = 1) (Acetic acid, glacial). Weighted average: 0.73 (Air = 1)		
Volatility	Not available.		
Odor Threshold	The highest known value is 0.48 ppm (Acetic acid, g	lacial)	
Water/Oil Dist. Coeff.	Not available.		
Ionicity (in Water)	Not available.		
<b>Dispersion Properties</b>	See solubility in water, diethyl ether, acetone.		
Solubility	Easily soluble in cold water, hot water. Soluble in diethyl ether, acetone.		

Section 10. Stability and Reactivity Data		
Stability	The product is stable.	
Instability Temperature	Not available.	
<b>Conditions of Instability</b>	Incompatible materials	
Incompatibility with various substances	Reactive with oxidizing agents. Slightly reactive to reactive with reducing agents, metals, acids, alkalis.	
Corrosivity	Not available	
Special Remarks on Reactivity	Not available	
Special Remarks on Corrosivity	Not available.	
Polymerization	Will not occur.	

Corrosivity	Not available.
Polymerization	Will not occur.
Section 11. Toxicolog	ical Information
Routes of Entry	Absorbed through skin. Eye contact. Inhalation. Ingestion.
<b>Toxicity to Animals</b>	Acute oral toxicity (LD50): 3310 mg/kg [Rat]. (Acetic acid, glacial). Acute dermal toxicity (LD50): 1060 mg/kg [Rabbit]. (Acetic acid, glacial).
<b>Chronic Effects on Humans</b>	MUTAGENIC EFFECTS: Mutagenic for bacteria and/or yeast. [Acetic acid, glacial].  May cause damage to the following organs kidneys, upper respiratory tract, skin, teeth.
Other Toxic Effects on Humans	Hazardous in case of skin contact (irritant), eye contact (irritant) Slightly hazardous in case of ingestion, of inhalation (lung irritant).
Special Remarks on Toxicity to Animals	Not available.
Special Remarks on Chronic Effects on Humans	May affect genetic material (mutagenic)
Special Remarks on other Toxic Effects on Humans	Acute Potential Health Effects Skin: causes skin imitation with redness and pain. Eyes: causes eye imitation with redness, pain and tearing. Inhalation: Causes respiratory tract and mucous membrane imitation with sneezing, wheezing, coughing, rhinitis, dyspnea, thoracic pain. This product contains Acetic acid. Inhalation of high concentrations of Acetic acid may affect behavior/central nervous system and cause central nervous system depression with headache, nausea, vomiting. Ingestion: Ingestion of large amounts may cause gastrointestinal (digestive) tract imitation with nausea, vomiting, diarrhea, abdominal pain. This product contains Acetic acid. Ingestion of large amounts of Acetic acid may affect urinary system - kidneys (albuminuria, hematuria, hemoglobinuria, anuria, uremia, nephrosis, acute renal failure, acute tubular necrosis), liver (impaired liver function), behavior (convulsions, giddines, muscular weakness). This product also contains Sodium Acetate, trihydrate. Ingestion of large amounts of Sodium Acetate may affect behavior/central nervous system, and urinary system. Chronic Potential Health Effects Ingestion and Inhalation: This product contains Acetic acid. Prolonged or repeated ingestion of large doses of Acetic acid may affect behavior, liver, kidneys, and metabolism (weight loss). Prolonged or repeated inhalation of Acetic acid may cause pharyngitis, chronic bronchitis, and may affect blood (changes in leukocyte count), and urinary system (kidney damage). Prolonged or repeated ingestion or inhalation of Acetic acid may also cause erosion of teeth. Skin: Prolonged or repeated skin contact with Acetic acid may cause imitation or dematitis, hyperkeratosis, cracking of the skin. Eyes Prolonged or repeated eye contact with Acetic acid may cause conjunctivitis

Section 12. Ecological Information				
Ecotoxicity	Not available.			
BOD5 and COD	Not available.			
Products of Biodegradation	Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.			
Toxicity of the Products of Biodegradation	The products of degradation are less toxic than the product itself.			
Special Remarks on the Products of Biodegradation	Not available.			

### Section 13. Disposal Considerations

Waste Disposal Waste must be disposed of in accordance with federal, state and local environmental control

regulations.

# Section 14. Transport Information DOT Classification Not a DOT controlled material (United States). Identification Not applicable. Special Provisions for Transport DOT (Pictograms)

### Section 15. Other Regulatory Information and Pictograms

Federal and State Regulations	Connecticut hazardous material survey.: Acetic acid, glacial Illinois toxic substances disclosure to employee act: Acetic acid, glacial Illinois chemical safety act: Acetic acid, glacial New York release reporting list: Acetic acid, glacial Rhode Island RTK hazardous substances: Acetic acid, glacial Pennsylvania RTK: Acetic acid, glacial Minnesota: Acetic acid, glacial Massachusetts RTK: Acetic acid, glacial Massachusetts spill list: Acetic acid, glacial New Jersey: Acetic acid, glacial New Jersey: Acetic acid, glacial Louisiana spill reporting: Acetic acid, glacial California Director's List of Hazardous Substances: Acetic acid, glacial TSCA 8(b) inventory: Water; Acetic acid, glacial CERCLA: Hazardous substances: Acetic acid, glacial: 5000 lbs (2268 kg);

California Proposition 65 Warnings	California prop. 65. This product contains the following ingredients for which the State of California has found to cause cancer which would require a warning under the statute: No products were found.
	California prop. 65: This product contains the following ingredients for which the State of California has found to cause birth defects which would require a warning under the statute: No products were found.
Other Regulations	OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).
Other Classifications	WHMIS (Canada) CLASS D-2B: Material causing other toxic effects (TOXIC).
	DSCL (FEC)

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## Sodium Acetate, 3 M Sterile Solution, pH 5.2 Page Number: 6 R36/37/38- Irritating to eyes, respiratory S26- In case of contact with eyes, rinse immediately with plenty of water and seek system and skin. medical advice. S37/39- Wear suitable gloves and eye/face protection. Health Hazard 2 **National Fire Protection** HMIS (U.S.A.) Flammability Association (U.S.A.) Fire Hazard 0 Health Reactivity 0 Personal Protection Specific hazard WHMIS (Canada) (Pictograms) DSCL (Europe) (Pictograms) TDG (Canada) (Pictograms) ADR (Europe) (Pictograms) **Protective Equipment** Gloves. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

# Section 16. Other Information MSDS Code S0099 References Not available. Other Special Considerations Validated by Sonia Owen on 4/15/2008. Verified by Sonia Owen. Printed 6/24/2008.

CALL (310) 516-8000

### **Notice to Reader**

All chemicals may pose unknown hazards and should be used with caution. This Material Safety Data Sheet (MSDS) applies only to the material as packaged. If this product is combined with other materials, deteriorates, or becomes contaminated, it may pose hazards not mentioned in this MSDS. It shall be the user's responsibility to develop proper methods of handling and personal protection based on the actual conditions of use. While this MSDS is based on technical data judged to be reliable, Spectrum Quality Products, Inc. assumes no responsibility for the completeness or accuracy of the information contained herein.