



Material Safety Data Sheet

NFPA	HMIS	Personal Protective Equipment
	Health Hazard 2 Fire Hazard 3	***
\checkmark	Reactivity 0	See Section 15.

Section 1. Chem	ical Product and Company Identification		Page Number: 1
Common Name/ Trade Name	Potassium Hydroxide. 0.1N Alcoholic Solution in Methanol	Catalog Number(s).	P-367
		CAS#	Mixture.
Manufacturer	SPECTRUM LABORATORY PRODUCTS INC.	RTECS	Not applicable.
	14422 S. SAN PEDRO STREET GARDENA, CA 90248	TSCA	TSCA 8(b) inventory: Potassium hydroxide; Methyl alcohol
Commercial Name(s)	Not available.	CI#	Not applicable.
Synonym	Not available.		
Chemical Name	Not applicable.		<u>5 EMERGENCY</u> C (24hr) 800-424-9300
Chemical Family	Alcohol. (Solvent.)	CALL (310) 5	16-8000
Chemical Formula	Not applicable.		
Supplier	SPECTRUM LABORATORY PRODUCTS INC. 14422 S. SAN PEDRO STREET GARDENA, CA 90248		

				Exposure Limits		
Name		CAS #	TWA (mg/m³)	STEL (mg/m³)	CEIL (mg/m³)	% by Weight
1) Potassium hydroxide 2) Methyl alcohol		1310-58-3 67-56-1	200	250	2	0.66 99.3
Toxicological Data on Ingredients Section 3. Hazards Io	Methyl alcohol: ORAL (LD50): DERMAL (LD50): VAPOR (LC50): Potassium hydroxid ORAL (LD50):		mg/kg [Rabbit]. ppm 4 hours [Rat].			
Potential Acute Health Effects	Hazardous in case of Slightly hazardous in may produce tissue contact may produce characterized by cou	case of skin contact (damage particularly burns. Inhalation	corrosive, permeate on mucous membr of the spray mist i	or), of eye contact anes of eyes, mo may produce seve	(corrosive). Lique buth and respirate are irritation of r	uid or spray mist ory tract. Skin espiratory tract,

Potassium Hydroxide. Solution in Methanol	0.1N Alcoholic Page Number: 2
Potential Chronic Health Effects	Slightly hazardous in case of skin contact (sensitizer). CARCINOGENIC EFFECTS : Not available. MUTAGENIC EFFECTS : Nutagenic for bacteria and/or yeast. [Methyl alcohol]. TERATOGENIC EFFECTS : Classified POSSIBLE for human [Methyl alcohol]. DEVELOPMENTAL TOXICITY : Not available. The substance is toxic to eyes. The substance may be toxic to blood, kidneys, liver, brain, peripheral nervous system, upper respiratory tract, skin, central nervous system (CNS), optic nerve. Repeated or prolonged exposure to the substance can produce target organs damage. Repeated or prolonged contact with spray mist may produce chronic eye irritation and severe skin irritation. Repeated or prolonged exposure to spray mist may produce respiratory tract irritation leading to frequent attacks of bronchial infection. Repeated exposure to a highly toxic material may produce general deterioration of health by an accumulation in one or many human organs.

Section 4. First Aid Measures

Eye Contact	Check for and remove any contact lenses. Immediately flush eyes with running water for at least 15 minutes, keeping eyelids open. Cold water may be used. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Cover the irritated skin with an emollient. Cold water may be used.Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.
Serious Skin Contact	Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.
Inhalation	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.
Serious Inhalation	Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. WARNING: It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled material is toxic, infectious or corrosive. Seek immediate medical attention.
Ingestion	If swallowed, do not induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention immediately.
Serious Ingestion	Not available.

Section 5. Fire and Explosion Data

Flammability of the Product	Flammable.
Auto-Ignition Temperature	The lowest known value is 464℃ (867.2年) (Methyl a lcohol).
Flash Points	The lowest known value is CLOSED CUP: 12℃ (53.6℉) . OPEN CUP: 16℃ (60.8℉). (Methyl alcohol)
Flammable Limits	The greatest known range is LOWER: 6% UPPER: 36.5% (Methyl alcohol)
Products of Combustion	These products are carbon oxides (CO, CO2). Some metallic oxides.
Fire Hazards in Presence of Various Substances	Highly flammable in presence of open flames and sparks, of heat. Non-flammable in presence of shocks, of oxidizing materials, of reducing materials, of combustible materials, of organic materials, of alkalis, of moisture.
Explosion Hazards in Presence of Various Substances	Risks of explosion of the product in presence of mechanical impact: Not available. Explosive in presence of open flames and sparks, of heat.
Fire Fighting Media and Instructions	Flammable liquid, soluble or dispersed in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use alcohol foam, water spray or fog.
Special Remarks on Fire Hazards	Explosive in the form of vapor when exposed to heat or flame. Vapor may travel considerable distance to source of ignition and flash back. When heated to decomposition, it emits acrid smoke and irritating fumes. CAUTION: MAY BURN WITH NEAR INVISIBLE FLAME (Methyl alcohol)
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Potassium Hydroxide Solution in Methano	
Special Remarks on Explosion Hazards	Forms an explosive mixture with air due to its low flash point. Explosive when mixed with Choroform + sodium methoxide and diethyl zinc. It boils violently and explodes. (Methyl alcohol)
Section 6. Accidental	Release Measures
Small Spill	Dilute with water and mop up, or absorb with an inert dry material and place in an appropriate waste disponent
Large Spill	Flammable liquid. Corrosive liquid. Poisonous liquid. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY ea sand or other non-combustible material. Do not get water inside container. Do not touch spilled material. U water spray curtain to divert vapor drift. Use water spray to reduce vapors. Prevent entry into sewe basements or confined areas; dike if needed. Call for assistance on disposal. Be careful that the product is present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.
Section 7. Handling a	nd Storage
Precautions	Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do ingest. Do not breathe gas/fumes/ vapor/spray. Never add water to this product. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidize agents, metals, acids.
Storage	Store in a segregated and approved area. Keep container in a cool, well-ventilated area. Keep container tig closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame).
Section 8. Exposure	Controls/Personal Protection
Engineering Controls	Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below the respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-statilocation.
Personal Protection	Face shield. Full suit. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Glov Boots.
Personal Protection in Case of a Large Spill	Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specia BEFORE handling this product.
Exposure Limits	Potassium hydroxide CEIL: 2 from ACGIH (TLV) [United States] [1999] Methyl alcohol TWA: 200 from OSHA (PEL) [United States] TWA: 200 STEL: 250 (ppm) from ACGIH (TLV) [United States] [1999] STEL: 250 from NIOSH [United States] TWA: 200 STEL: 250 (ppm) from NIOSH SKIN TWA: 200 STEL: 250 (ppm) from NIOSH SKIN TWA: 200 STEL: 250 (ppm) [Canada]

Consult local authorities for acceptable exposure limits.

Section 9. Physical and Chemical Properties

Physical state and appearance	Liquid.	Odor	Alcohol like.
		Taste	Not available.
Molecular Weight	Not applicable.	Color	Clear Colorless.
pH (1% soln/water)	Not available.	Color	
Boiling Point	The lowest known value is 64.5 °C (148.1°F) (Methyl	alcohol).	
Melting Point	May start to solidify at -97.8 $\ensuremath{^\circ}$ (-144 $\ensuremath{^\circ}$) based on $\ensuremath{^\circ}$	lata for: M	ethyl alcohol.
Critical Temperature	The lowest known value is 240 ${\rm C}$ (464 ${\rm F}$) (Methyl ald	ohol).	
Specific Gravity	The only known value is 0.7915 (Water = 1) (Methy	'l alcohol).	
Vapor Pressure	The highest known value is 12.3 kPa (@ 20°C) (Me	thy I alcoho	ol).
Vapor Density	The highest known value is 1.11 (Air = 1) (Methyl	alcohol).	

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Potassium Hydroxide Solution in Methanol	
Volatility	Not available.
Odor Threshold	The highest known value is 100 ppm (Methyl alcohol)
Water/Oil Dist. Coeff.	Not available.
Ionicity (in Water)	Non-ionic.
Dispersion Properties	See solubility in water.
Solubility	Easily soluble in cold water, hot water. Insoluble in diethyl ether.
Section 10. Stability a	and Reactivity Data
Stability	The product is stable.
Instability Temperature	Not available.
Conditions of Instability	Heat, ignition sources, incompatible materials
Incompatibility with various substances	Reactive with oxidizing agents, metals, acids.
Corrosivity	Corrosive in presence of aluminum, of zinc. Non-corrosive in presence of glass, of stainless steel(316).
Special Remarks on Reactivity	Can react vigorously with oxidizers. Violent reaction with alkyl aluminum salts, acetyl bromide, chloroform sodium methoxide, chromic anhydride, cyanuirc chlorite, lead perchlorate, phosphorous trioxide, nitric acid. Exothermic reaction with sodium hydroxide + chloroform. Incompatible with beryllium dihydride, metals (potassium and magnesium), oxidants (barium perchlorate, bromir sodium hypochlorite, chlorine, hydrogen peroxide), potassium tert-butoxide, carbon tetrachloride, alkali metals (aluminum, potassium magnesium, zinc), and dichlormethane. Rapid autocatalytic dissolution of aluminum, magnesium or zinc in 9:1 methanol + carbon tetrachloride sufficiently vigorous to be rated as potentially hazardous. May attack some plastics, rubber, and coatings. (Methyl alcohol)
Special Remarks on Corrosivity	When wet, attacks metals such as aluminum, tin, lead, and zinc, producing flammable hydrogen gas. Severe corrosive effect on brass and bronze. (Potassium hydroxide)
Polymerization	Will not occur.
Section 11. Toxicolog	yical Information
Routes of Entry	Absorbed through skin. Eye contact. Inhalation. Ingestion.
Toxicity to Animals	Acute oral toxicity (LD50): 5628 mg/kg [Rat.]. (Methyl alcohol). Acute dermal toxicity (LD50): 15800 mg/kg [Rabbit.]. (Methyl alcohol).
Chronic Effects on Humans	MUTAGENIC EFFECTS: Mutagenic for bacteria and/or yeast. [Methyl alcohol]. TERATOGENIC EFFECTS: Classified POSSIBLE for human [Methyl alcohol]. Contains material which may cause damage to the following organs: blood, kidneys, liver, brain, periphe nervous system, upper respiratory tract, skin, central nervous system (CNS), optic nerve.
Other Toxic Effects on Humans	Hazardous in case of skin contact (irritant), of ingestion, of inhalation (lung irritant). Slightly hazardous in case of skin contact (permeator).
Special Remarks on Toxicity to Animals	Not available.
	Passes through the placental barrier.
Special Remarks on Chronic Effects on Humans	May affect genetic material. May cause birth defects and adverse reproductive effects(paternal and maternal effects and fetotoxicity) bas on animal studies. (Methyl alcohol)

Potassium Hydroxide. 0.1N Alcoholic Solution in Methanol	Page Number: 5
Solution in Methanol Acute Potential Health effects: May cause eye and skin irritation. Methanol can be absorbed throu include visual disturbances. Eye contact may also cause conjunctivit Inhalation: May cause respiratory tract irritation with coughing a nervous system/peripherial nervous system, gastrointestinal tract, /cardiovascular system (bradycardia, tachydardia). May also cause which may include reduced reactivity/and or increased sensitivity t and blindness. Ingestion: May be harmful and affect eyes (cause significant v swallowed. May cause gastrointestinal tract irritation with abdominal or constipation. May affect behavior/central nervous system/perip dizziness, delirum, confusion, restlessness, giddiness, back pain, spastic paralysis, muscle contraction, ataxia, seizures, unconcio metabolism, respiration (dyspnea, apnea, hyperventilation, pulmonary urinary system (kidneys - renal failure, hematuria), endocrine hyperglycemia)), cardiovascular system (tachycardia, bradycardia cause metabolic acidosis. Narcotic. Chronic Potential Effects: Prolonged or repeated exposure by inhalation or ingestion will have ingestion. Methanol is very slowly eliminated from the body. Because of this sl as a cumulative poison. Though a single exposure may cause n accumulation of harmful amounts Prolonged or repeated skin contact may cause deffating dermatitis w	tis and possible corneal damage. and wheezing. May affect behavior/central r, respiration, lungs, and blood, and heart e metabolic acidosis and severe visual effects to light, blurred, double/and or snowy vision, visual disturbances including blindness) if pain, fatigue, nausea, vomiting, and diarrhea pheral nervous system (general anesthetic, headache, muscle weakness, somnolence, busness, coma), brain, blood(leukocytosis), y edema, coughing, respiratory failure), liver, e system (spleen, pancreas (pancreatitis, a, cardiac failure, hypotension). May also effects similar to those of acute inhalation or slow elimination, methanol should be regarded no effect, daily exposures may result in the
Section 12. Ecological Information	

Ecotoxicity	Not available.
BOD5 and COD	Not available.
Products of Biodegradation	Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.
Toxicity of the Products of Biodegradation	The products of degradation are less toxic than the product itself.
Special Remarks on the Products of Biodegradation	Methanol in water is rapidly biodegraded and volatilized. Aquatic hydrolysis, oxidation, photolysis, adsorption to sediment, and bioconcentration are not significant fate processes. The half-life of methanol in surfact water ranges from 24 hrs. to 168 hrs. Based on its vapor pressure, methanol exists almost entirely in the vapor phase in the ambient atmosphere. It is degraded by reaction with photochemically produced hydroxyl radicals and has an estimated half-life of 17.8 days. Methanol is physically removed from air by rain due to its solubility. Methanol can react with NO2 in pollulted to form methyl nitrate. The half-life of methanol in air ranges from 71 hrs. (3 days) to 713 hrs. (29.7 days) based on photooxidation half-life in air. (Methyl alcohol)

Section 13. Disposal Considerations

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Waste Disposal

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

DOT Classification	CLASS 3: Flammable liquid.
Identification	UNNA: 1230 : Methanol, solution PG: II
Special Provisions for Transport	Not available.

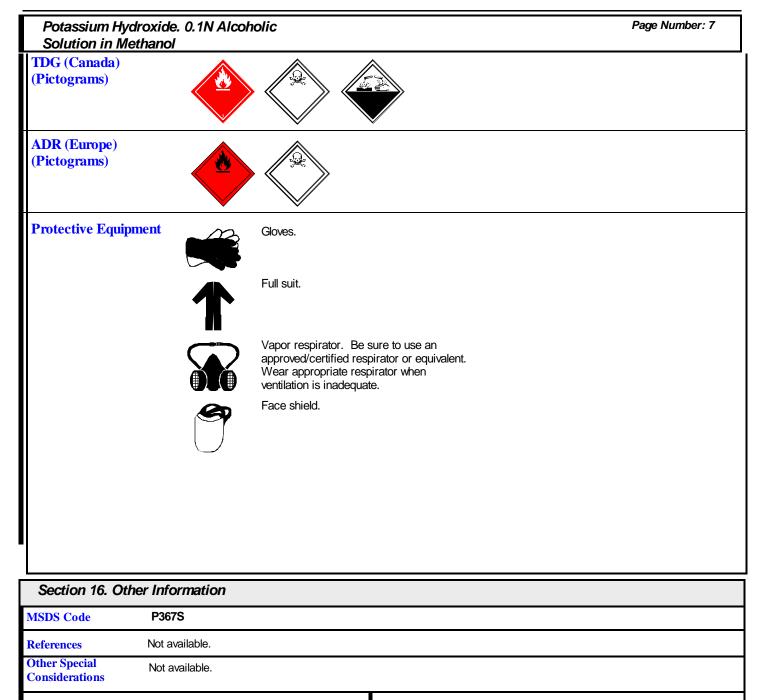
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DOT (Pictograms)



Section 15. Other Regulatory Information and Pictograms				
Federal and State Regulations	Connecticut hazardous material survey.: Methyl alcohol Illinois toxic substances disclosure to employee act: Methyl alcohol Illinois chemical safety act: Methyl alcohol New York release reporting list: Potassium hydroxide; Methyl alcohol Rhode Island RTK hazardous substances: Methyl alcohol Pennsylvania RTK: Potassium hydroxide; Methyl alcohol Florida: Potassium hydroxide; Methyl alcohol Minnesota: Potassium hydroxide; Methyl alcohol Massachusetts RTK: Potassium hydroxide; Methyl alcohol Massachusetts spill list: Methyl alcohol New Jersey: Potassium hydroxide; Methyl alcohol New Jersey spill list: Methyl alcohol Louisiana spill reporting: Methyl alcohol TSCA 8(b) inventory: Potassium hydroxide; Methyl alcohol SARA 313 toxic chemical notification and release reporting: Methyl alcohol 99.34% CERCLA: Hazardous substances.: Potassium hydroxide: 1000 lbs. (453.6 kg); Methyl alcohol: 5000 lbs. (2268 kg);			
California Proposition 65 Warnings	California prop. 65. This product contains the following ingredients for which the State of California has found to cause cancer which would require a warning under the statute: No products were found. California prop. 65: This product contains the following ingredients for which the State of California has found to cause birth defects which would require a warning under the statute: No products were found.			
Other Regulations	OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).			
Other Classifications	WHMIS (Canada)	WHMIS (Canada) CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). CLASS D-1A: Material causing immediate and serious toxic effects (VERY TOXIC). CLASS D-2B: Material causing other toxic effects (TOXIC). CLASS E: Corrosive liquid.		
	DSCL (EEC)	R11- Highly flammable. R23/24/25- Toxic by inhalation, in contact with skin and if swallowed. R36/37/38- Irritating to eyes, respiratory system and skin.	 S7- Keep container tightly closed. S16- Keep away from sources of ignition - No smoking. S36/37/39- Wear suitable protective clothing, gloves and eye/face protection. S45- In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible). 	
HMIS (U.S.A.)	Health Hazard Fire Hazard Reactivity Personal Protection	2 National Fire Protection 3 Association (U.S.A.)	Health Flammability Reactivity Specific hazard	
WHMIS (Canada) (Pictograms)				
DSCL (Europe) (Pictograms)				

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Validated by Sonia Owen on 3/20/2008.

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CALL (310) 516-8000

Notice to Reader

All chemicals may pose unknown hazards and should be used with caution. This Material Safety Data Sheet (MSDS) applies only to the material as packaged. If this product is combined with other materials, deteriorates, or becomes contaminated, it may pose hazards not mentioned in this MSDS. It shall be the user's responsibility to develop proper methods of handling and personal protection based on the actual conditions of use. While this MSDS is based on technical data judged to be reliable, Spectrum Quality Products, Inc. assumes no responsibility for the completeness or accuracy of the information contained herein.