



Material Safety Data Sheet

NFPA	HMIS	Personal Protective Equipment
	Health Hazard 1 Fire Hazard 0	
	Reactivity 0	See Section 15.

Section 1. Chem	ical Product and Company Identification		Page Number: 1
Common Name/ Trade Name	Nitrogen Standard, Solution, 1 ml = 0.1 mg N.	Catalog Number(s).	N-151
	IN.	CAS#	Mixture.
Manufacturer	SPECTRUM LABORATORY PRODUCTS INC.	RTECS	Not applicable.
	14422 S. SAN PEDRO STREET GARDENA, CA 90248	TSCA	TSCA 8(b) inventory: Ammonium chloride; Water
Commercial Name(s)	Not available.	CI#	Not applicable.
Synonym	Not available.		
Chemical Name	Not applicable.		<u>EMERGENCY</u> (24hr) 800-424-9300
Chemical Family	Chloride salt. (Salt.)	CALL (310) 5 ²	16-8000
Chemical Formula	Not applicable.		
Supplier	SPECTRUM LABORATORY PRODUCTS INC. 14422 S. SAN PEDRO STREET GARDENA, CA 90248		

Section 2.Composition and Information on Ingredients						
				Exposure Limits		
Name		CAS #	TWA (mg/m ³)	STEL (mg/m ³)	CEIL (mg/m ³)	% by Weight
1) Ammonium chloride 2) Water	12125-02-9 7732-18-5					
Toxicological Data on Ingredients Ammonium chloride: ORAL (LD50): Acute: 1650 mg/kg [Rat.]. 1300 mg/kg [Mouse].						
Section 3. Hazards Identification						
Potential Acute Health Effects	Slightly hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion. Non-hazardous in case of inhalation.					
Potential Chronic Health Effects	CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available.					

Section 4. First Aid Measures Eye Contact Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention if irritation occurs. **Skin Contact** Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if irritation develops. Cold water may be used. Serious Skin Contact Not available. Inhalation If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention if symptoms appear. **Serious Inhalation** Not available. Ingestion Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband. Serious Ingestion Not available.

Section 5. Fire and E	Section 5. Fire and Explosion Data					
Flammability of the Product	Non-flammable.					
Auto-Ignition Temperature	Not applicable.					
Flash Points	Not applicable.					
Flammable Limits	Not applicable.					
Products of Combustion	Not available.					
Fire Hazards in Presence of Various Substances	Not applicable.					
Explosion Hazards in Presence of Various Substances	Non-explosive in presence of open flames and sparks, of shocks.					
Fire Fighting Media and Instructions	Not applicable.					
Special Remarks on Fire Hazards	Not available.					
Special Remarks on Explosion Hazards	Explosive reaction between bromine trifluoride and ammonium halides. (Ammonium chloride)					
Section 6. Accidental	Release Measures					
Small Spill	Dilute with water and mop up, or absorb with an inert dry material and place in an appropriate waste disposal container. Finish cleaning by spreading water on the contaminated surface and dispose of according to local and regional authority requirements.					

e Spill	Absorb with an inert material and put the spilled material in an appropriate waste disposal. Finish cleaning by
	spreading water on the contaminated surface and allow to evacuate through the sanitary system.

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Section 7. Handling a	Section 7. Handling and Storage						
Precautions	No specific safety phrase has been found applicable for this product.						
Storage	Keep container tightly closed. Keep container in a cool, well-ventilated area.						
Section 8. Exposure	Controls/Personal Protection						
Engineering Controls	Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value.						
Personal Protection	Safety glasses. Synthetic apron. Gloves (impervio	us).					
Personal Protection in Case of a Large Spill	Splash goggles. Full suit. Boots. Gloves. Sug specialist BEFORE handling this product.	gested pr	otective clothing might not be sufficient; consult a				
Exposure Limits	Ammonium Chloride: TWA: 10 STEL: 20 (mg/m ³) from ACGIH (TLV) [United States] Inhalation TWA: 10 STEL: 20 (mg/m ³) [United Kingdom (UK)] Inhalation TWA: 10 STEL: 20 (mg/m ³) from NIOSH [United States] Inhalation TWA: 10 STEL: 20 (mg/m ³) from OSHA (PEL) [United States]						
	Consult local authorities for acceptable exposure lim	nits.					
Section 9. Physical and Chemical Properties							
Physical state and appearance	Liquid.	Odor	Not available.				
Molecular Weight	Not applicable.	Taste	Not available.				
pH (1% soln/water)	Neutral.	Color Clear Colorless.					
Boiling Point	The lowest known value is 100°C (212°F) (Water).						
Melting Point	Not available.						
Critical Temperature	Not available.						
Specific Gravity	The only known value is 1 (Water = 1) (Water).						
Vapor Pressure	The highest known value is 2.3 kPa (@ 20°C) (Wat	er).					
Vapor Density	The highest known value is 0.62 (Air = 1) (Water).	1					
Volatility	Not available.						
Odor Threshold	Not available.						
Water/Oil Dist. Coeff.	Not available.						
Ionicity (in Water)	Not available.						
Dispersion Properties	See solubility in water, methanol.						
Solubility	Easily soluble in cold water, hot water. Soluble in methanol. Insoluble in diethyl ether, acetone.						

Nitrogen Standard, Solution, 1 ml = 0.1 mg

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Section 10. Stability	and Reactivity Data
Stability	The product is stable.
Instability Temperature	Not available.
Conditions of Instability	Incompatible materials
Incompatibility with various substances	Not available.
Corrosivity	Not available
Special Remarks on Reactivity	Incompatible with lead and silver salts. It can react violently with ammonium nitrate and potassium chlorate. Also incompatible with bromine trifluoride, ammonium halides, bromine pentafluoride, alkalis and their carbonates. At fire temperature, ammonium chloride may dissociate into ammonia and hydrogen chloride. Hygroscopic; keep container tightly closed. (Ammonium chloride)
Special Remarks on Corrosivity	Not available.
Polymerization	Will not occur.
Section 11. Toxicolo	ogical Information
Routes of Entry	Skin contact. Eye contact.
Toxicity to Animals	Ammonium chloride: ORAL (LD50): Acute: 1650 mg/kg [Rat.]. 1300 mg/kg [Mouse].
Chronic Effects on Humans	Not available.
Other Toxic Effects on Humans	Slightly hazardous in case of skin contact (irritant), of ingestion. Non-hazardous in case of inhalation.
Special Remarks on Toxicity to Animals	Lowest Published Lethal Dose: LDL [Human Infant] - Route: Oral; Dose: 2000 mg/kg (Ammonium chloride)
Special Remarks on Chronic Effects on Humans	May affect genetic material (mutagenic) Animal: passes through the placental barrier. (Ammonium chloride)
Special Remarks on other Toxic Effects on Humans	Acute Potential Health Effects: Skin: Low Hazard. May cause irritation, particularly for those individuals with sensitive skin. Eyes: Low Hazard. May cause eye irritation. Inhalation: Not expected to be an inhalation hazard. Ingestion: Very low hazard for normal handling. May cause gastrointestinal tract disturbances. May cause nausea, vomiting. The product contains Ammonium Chloride. Ingestion of large amounts of Ammonium Chloride may affect behavior/central nervous system (headache, somnolence, confusion, drowsiness, tremor, convulsions, coma), eyes (Mydriasis), cardiovascular system (bradycardia), respiration (respiratory stimulation, apnea, hyperventilation, pulmonary edema). May cause serious metabolic acidosis with hypokalemia. Transient hyperglycemia and glycosuria may also occur. Chronic Potential Health Effects: Skin: Prolonged or repeated contact with Ammonium Chloride may cause dermatitis, an allergic reaction. Ingestion: Prolonged or repeated ingestion of Ammonium Chloride may affect metabolism (anorexia, metabolic acidosis) and urinary system (enlargement of kidneys).

Section 12. Ecologie	cal Information
Ecotoxicity	Not available.
BOD5 and COD	Not available.
Products of Biodegradation	Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.
Toxicity of the Products of Biodegradation	Not available.
Special Remarks on the Products of Biodegradation	Not available.

Section 13. Disposal Considerations

Waste Disposal Waste must be disposed of in accordance with federal, state and local environmental control regulations.

DOT Classification	Not a DOT controlled material (United States).	
Identification	Not applicable.	
Special Provisions for Transport	Not applicable.	
DOT (Pictograms)		

Federal and State Regulations	Illinois toxic substances disclosure to employee act: Ammonium chloride Illinois chemical safety act: Ammonium chloride New York release reporting list: Ammonium chloride Rhode Island RTK hazardous substances: Ammonium chloride Pennsylvania RTK: Ammonium chloride Minnesota: Ammonium chloride Massachusetts RTK: Ammonium chloride Massachusetts spill list: Ammonium chloride New Jersey: Ammonium chloride New Jersey spill list: Ammonium chloride Louisiana spill reporting: Ammonium chloride California Director's List of Hazardous Substances: Ammonium Chloride TSCA 8(b) inventory: Ammonium chloride; Water CERCLA: Hazardous substances.: Ammonium chloride: 5000 lbs. (2268 kg);
California Proposition 65 Warnings	California prop. 65: This product contains the following ingredients for which the State of California has found to cause cancer which would require a warning under the statute: No products were found. California prop. 65: This product contains the following ingredients for which the State of California has found to cause birth defects which would require a warning under the statute: No products were found.
Other Regulations	

Nitrogen Standard, N.	Solution, 1 ml = 0	0.1 mg				Page Number: 6
	235-186-4). Canada: Listed on Na Japan: Listed on Na Korea: Listed on Na Philippines: Listed on Australia: Listed on For Water: EINECS: This produ 231-791-2).	uct is on the Canadian D Itional Inver ational Inver tional Inver on National AICS. uct is on the Canadian D	ntory (ENCS). htory (KECI).			
Other Classifications	WHMIS (Canada)	Not con	trolled under WHMIS (Canada).			
	DSCL (EEC)		duct is not classified according J regulations.	Not appli	cable.	
HMIS (U.S.A.)	Health Hazard Fire Hazard Reactivity Personal Protection	1 0 0 C	National Fire Protection Association (U.S.A.)	Health	000	Flammability Reactivity Specific hazard
WHMIS (Canada) (Pictograms)						
DSCL (Europe) (Pictograms)						
TDG (Canada) (Pictograms)	\bigotimes					
ADR (Europe) (Pictograms)						
Protective Equipment	Glov	es (imperv	ious).			
	Synt	hetic apror	1.			
		applicable. ty glasses.				
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Section 16. Other Information					
MSDS Code	N151S				
References	Not available.				
Other Special Considerations	Not available.				
Validated by Sonia	Owen on 8/11/2006.	Verified by Sonia Owen. Printed 9/12/2006.			
CALL (310) 516-80	00	-			

Notice to Reader

All chemicals may pose unknown hazards and should be used with caution. This Material Safety Data Sheet (MSDS) applies only to the material as packaged. If this product is combined with other materials, deteriorates, or becomes contaminated, it may pose hazards not mentioned in this MSDS. It shall be the user's responsibility to develop proper methods of handling and personal protection based on the actual conditions of use. While this MSDS is based on technical data judged to be reliable, Spectrum Quality Products, Inc. assumes no responsibility for the completeness or accuracy of the information contained herein.