



Material Safety Data Sheet



Section 1. Chemical Product and Company Identification			Page Number: 1	
Common Name/ Trade Name	Nitric Acid, 0.1N		Catalog Number(s).	N-145
			CAS#	Mixture.
Manufacturer	SPECTRUM LABORATORY PRODUCTS IN	C.	RTECS	Not applicable.
	14422 S. SAN PEDRO STREET GARDENA, CA 90248		TSCA	TSCA 8(b) inventory: Water; Nitric acid, fuming
Commercial Name(s)	Not available.		CI#	Not applicable.
Synonym	Not available.			
Chemical Name	Not applicable.		<u>CHEMTREC</u>	(24hr) 800-424-9300
Chemical Family	(Acid.)		CALL (310) 5	16-8000
Chemical Formula	Not applicable.			
Supplier	SPECTRUM LABORATORY PRODUCTS INC. 14422 S. SAN PEDRO STREET GARDENA, CA 90248			

Section 2.Composition and Information on Ingredients						
				Exposure Limits		
Name		CAS #	TWA (mg/m ³)	STEL (mg/m ³)	CEIL (mg/m ³)	% by Weight
1) Water 2) Nitric acid, fuming		7732-18-5 7697-37-2	2	4		99.37 0.63
Toxicological Data on Ingredients	Nitric acid, fuming: VAPOR (LC50): Acute: 244 ppm 0.5 hours [Rat]. 344 ppm 0.5 hours [Rat].					
Section 3. Hazards	Identification					
Potential Acute Health Effects	Hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion. Slightly hazardous in case of skin contact (permeator), of inhalation.					
Potential Chronic Health Effects	CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available.					

ffects	MUTAGENIC EFFECTS: Not available.
	TERATOGENIC EFFECTS: Not available.
	DEVELOPMENTAL TOXICITY: Not available.
	The substance may be toxic to lungs, mucous membranes, upper respiratory tract, skin, eyes, teeth.
	Repeated or prolonged exposure to the substance can produce target organs damage.

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Section 4. First Aid N	leasures
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Cover the irritated skin with an emollient. Cold water may be used.Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.
Serious Skin Contact	Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek medical attention.
Inhalation	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.
Serious Inhalation	Not available.
Ingestion	Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.
Serious Ingestion	Not available.
Section 5. Fire and E	xplosion Data
Flammability of the Product	Non-flammable.
Auto-Ignition Temperature	Not applicable.
Flash Points	Not applicable.
Flammable Limits	Not applicable.
Products of Combustion	Not available.
Fire Hazards in Presence of Various Substances	Not applicable.
Explosion Hazards in Presence of Various Substances	Non-explosive in presence of open flames and sparks, of shocks.
Fire Fighting Media and Instructions	Not applicable.
Special Remarks on Fire Hazards	Not available.
Special Remarks on Explosion Hazards	Reacts exlposively with metallic powders, carbides, cyanides, sulfides, alkalies and turpentine. Can react explosively with many reducing agents. Arsine, phosphine, tetraborane all oxidized explosively in presence of fuming nitric acid. Cesium and rubidium acetylides explode in contact with nitric acid. Explosive reaction with Nitric Acid + Nitrobenzene + water. Detonation with Nitric Acid + 4-Methylcyclohexane. The addition of warm fuming nitric acid to phosphine causes explosion. Addition of water to nitration mixture diluted with an equal volume of water can cause a low order explosion. Cyclopentadiene reacts explosively with fuming nitric acid. Mixtures of fuming nitric acid and acetonitrile are high explosives. (Nitric acid, fuming)

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Section 6. Accidental	Release Measures			
Small Spill	Dilute with water and mop up, or absorb with an inert dry material and place in an appropriate waste disposal container. If necessary: Neutralize the residue with a dilute solution of sodium carbonate. Finish cleaning by spreading water on the contaminated surface and dispose of according to local and regional authority requirements.			
Large Spill	Absorb with an inert material and put the spilled material in an appropriate waste disposal. Neutralize the residue with a dilute solution of sodium carbonate. Finish cleaning by spreading water on the contaminated surface and allow to evacuate through the sanitary system. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.			
Section 7. Handling a	and Storage			
Precautions	Keep container dry. Do not breathe gas/fumes/ unwell, seek medical attention and show the labe	vapor/sp I when po	bray. Never add water to this product. If you feel bossible. Avoid contact with skin and eyes.	
Storage	Keep container tightly closed. Keep container in	a cool, w	ell-ventilated area.	
Section 8. Exposure	Controls/Personal Protection			
Engineering Controls	Provide exhaust ventilation or other engineerin below their respective threshold limit value. Ens to the work-station location.	g contro ure that e	Is to keep the airborne concentrations of vapors eyewash stations and safety showers are proximal	
Personal Protection	Splash goggles. Synthetic apron. Gloves (impervious). Respiratory protection is not necessary for normal handling. Good room ventilation or use of local exhaust (fume hood) is sufficient. Use a vapor respirator under conditions where exposure to the substance is apparent (e.g. generation of high concentrations of mist or vapor, inadequate ventilation, development of respiratory tract irritation), and engineering controls are not feasible. Be sure to use an approved/certified respirator or equivalent.			
Personal Protection in Case of a Large Spill	Splash goggles. Full suit. Boots. Gloves. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.			
Exposure Limits	Nitric acid, fuming TWA: 2 STEL: 4 (ppm) from ACGIH (TLV) [United States] [1999] TWA: 2 STEL: 4 (ppm) [Australia] TWA: 2 STEL: 4 from NIOSH TWA: 5 STEL: 10 (mg/m ³) from NIOSH TWA: 2 STEL: 4 (ppm) from OSHA (PEL) [United States] TWA: 5 STEL: 10 (mg/m ³) from OSHA (PEL) [United States] Consult local authorities for acceptable exposure limits.			
Section 9. Physical a	nd Chemical Properties			
Physical state and appearance	Liquid.	Odor	Not available.	
Molecular Weight	Not applicable	Taste	Not available.	
nH (1% soln/water)	Acidic	Color	Not available.	
Boiling Point				
Melting Point	Not available			
Critical Temperature	Not available			
Specific Gravity	The only known value is 1 (Water = 1) (Water).			
Vapor Pressure	The highest known value is 2.3 kPa (@ 20°C) (Water).			
Vapor Density	The highest known value is 0.62 (Air = 1) (Water).			
Volatility	Not available.			
Odor Threshold	Not available.			
Water/Oil Dist. Coeff.	Not available.			
Ionicity (in Water)	Not available.			
Dispersion Properties	See solubility in water, diethyl ether.			
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Solubility	Easily soluble in cold water, hot water. Soluble in diethyl ether.
Section 10. Stability	and Reactivity Data
Stability	The product is stable.
Instability Temperature	Not available.
Conditions of Instability	Incompatible materials
Incompatibility with various substances	Slightly reactive to reactive with reducing agents, combustible materials, organic materials, metals, alkalis.
Corrosivity	Corrosive in presence of aluminum, of copper. Non-corrosive in presence of glass, of stainless steel(304), of stainless steel(316).
Special Remarks on Reactivity	A strong oxidizer. Reacts violently with alcohol, organic material, turpene, charcoal. Violent reaction with Nitric acid + Acetone and Sulfuric acid. Incompatible with combustible materials, metallic powders, hydrogen sulfide, carbides, aldehydes, cyanides, chromic acid, hydrogen sulfide, metals, metal powders, organic solvents, acetic acid, alcohols. Nitric Acid will react with water or steam to produce heat and toxic, corrosive and flammable vapors. (Nitric acid, fuming)
Special Remarks on Corrosivity	Not available.
Polymerization	Will not occur.
Section 11. Toxicolog	gical Information
Routes of Entry	Absorbed through skin. Eye contact. Ingestion.
Toxicity to Animals	LD50: Not available. LC50: Not available.
Chronic Effects on Humans	Contains material which may cause damage to the following organs: lungs, mucous membranes, upper respiratory tract, skin, eyes, teeth.
Other Toxic Effects on Humans	Hazardous in case of skin contact (irritant), of ingestion. Slightly hazardous in case of skin contact (permeator), of inhalation.
Special Remarks on Toxicity to Animals	LDL - Lowest Published Lethal Dose [Human] - Route: Oral; Dose: 430 mg/kg (Nitric acid, fuming)
Special Remarks on Chronic Effects on Humans	May cause adverse reproductive effects (effects on newborn and fetotoxicity) based on animal data. (Nitric acid, fuming)
Special Remarks on other Toxic Effects on Humans	Acute Potential Health Effects: Skin: Causes skin irritation. It may cause a yellow-brown discoloratiion of the skin. Eyes: Causes eye irritation. Inhalation: Inhalation of mist or vapor may cause respiratory tract irritation. Ingestion: Can cause gastrointestinal/digestive tract irritation with nausea, vomiting, diarrhea. Chronic Potential Health Effects: Repeated inhalation may produce changes in pulmonary function and/or chronic bronchitis. It may also affect behavior (headache, dizziness, drowsiness, muscle contaction or spasticity, weakness, loss of coordinaton, mental confusion), and urinary system (kidney faillure, decreased urinary output after several hours of uncorrected circulatory collapse). Repeated exposure may cause discoloration and/or errosion of teeth (dental enamel). Eye irritation and respiratory tract signs and symptoms resembling those of frequent upper respiratory viral infections have been associated with chronic nitric acid exposure.

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Section 12. Ecological Information		
Ecotoxicity	Not available.	
BOD5 and COD	Not available.	
Products of Biodegradation	Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.	
Toxicity of the Products of Biodegradation	The product itself and its products of degradation are not toxic.	
Special Remarks on the Products of Biodegradation	Not available.	

Section 13. Disposal Considerations

Waste Disposal

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

Section 14. Transport Information		
DOT Classification	Not a DOT controlled material (United States).	
Identification	Not applicable.	
Special Provisions for Transport	Marine Pollutant	
DOT (Pictograms)		

Section 15. Other Regulatory Information and Pictograms		
Federal and State Regulations	Connecticut hazardous material survey.: Nitric acid, 70% Illinois toxic substances disclosure to employee act: Nitric acid, 70% Illinois chemical safety act: Nitric acid, 70% New York release reporting list: Nitric acid, 70% Rhode Island RTK hazardous substances: Nitric acid, 70% Pennsylvania RTK: Nitric acid, 70% Florida: Nitric acid, 70% Minnesota: Nitric acid, 70% Massachusetts RTK: Nitric acid, 70% Massachusetts spill list: Nitric acid, 70% New Jersey: Nitric acid, 70% New Jersey spill list: Nitric acid, 70% New Jersey toxic catastrophe prevention act: Nitric acid, 70% Louisiana RTK reporting list: Nitric acid, 70% TSCA 8(b) inventory: Water; Nitric acid, fuming SARA 302/304/311/312 extremely hazardous substances: Nitric acid, 70% CERCLA: Hazardous substances:: Nitric acid, 70%: 1000 lbs. (453.6 kg);	
California Proposition 65 Warnings	California prop. 65: This product contains the following ingredients for which the State of California has found to cause cancer which would require a warning under the statute: No products were found. California prop. 65: This product contains the following ingredients for which the State of California has found to cause birth defects which would require a warning under the statute: No products were found.	
Other Regulations	OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).	
Other Classifications	WHMIS (Canada) CLASS D-2B: Material causing other toxic effects (TOXIC).	
	DSCL (EEC)	

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Section 16. Other Information			
MSDS Code	N145S		
References	Not available.		
Other Special Considerations	Not available.		
Validated by Sonia Owen on 2/18/2010.		Verified by Sonia Owen. Printed 2/18/2010.	
CALL (310) 516-8000			

Notice to Reader

All chemicals may pose unknown hazards and should be used with caution. This Material Safety Data Sheet (MSDS) applies only to the material as packaged. If this product is combined with other materials, deteriorates, or becomes contaminated, it may pose hazards not mentioned in this MSDS. It shall be the user's responsibility to develop proper methods of handling and personal protection based on the actual conditions of use. While this MSDS is based on technical data judged to be reliable, Spectrum Quality Products, Inc. assumes no responsibility for the completeness or accuracy of the information contained herein.